

Cytosinium hydrogen chloranilate monohydrate

Kazuma Gotoh, Rie Ishikawa and
Hiroyuki Ishida*Department of Chemistry, Faculty of Science,
Okayama University, Okayama 700-8530,
JapanCorrespondence e-mail:
ishidah@cc.okayama-u.ac.jp

Key indicators

Single-crystal X-ray study
 $T = 100$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.028
 wR factor = 0.073
Data-to-parameter ratio = 15.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

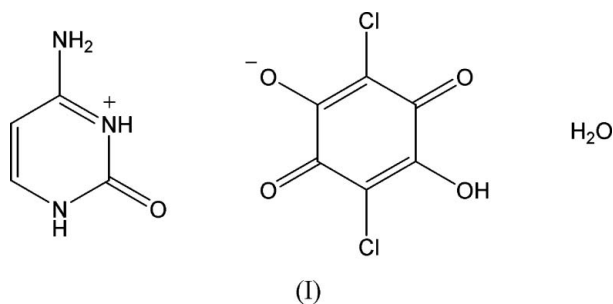
The title compound, $\text{C}_4\text{H}_6\text{N}_3\text{O}^+ \cdot \text{C}_6\text{HCl}_2\text{O}_4^- \cdot \text{H}_2\text{O}$, is a monohydrate salt of chloranilic acid (2,5-dichloro-3,6-dihydroxy-1,4-benzoquinone) with cytosine [4-aminopyrimidine-2(1*H*)-one]. In the crystal structure, the cytosinium cation, the hydrogen chloranilate anion and the water molecule are held together through $\text{O}-\text{H} \cdots \text{O}$, $\text{N}-\text{H} \cdots \text{O}$, $\text{N}-\text{H} \cdots \text{Cl}$ and $\text{C}-\text{H} \cdots \text{O}$ hydrogen bonds, forming a molecular tape. The tapes are further linked by $\text{O}-\text{H} \cdots \text{Cl}$ and $\text{C}-\text{H} \cdots \text{Cl}$ hydrogen bonds, forming a three-dimensional hydrogen-bond network.

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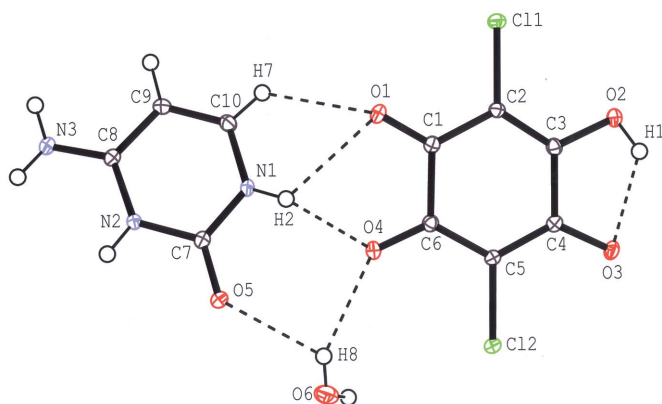
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Comment

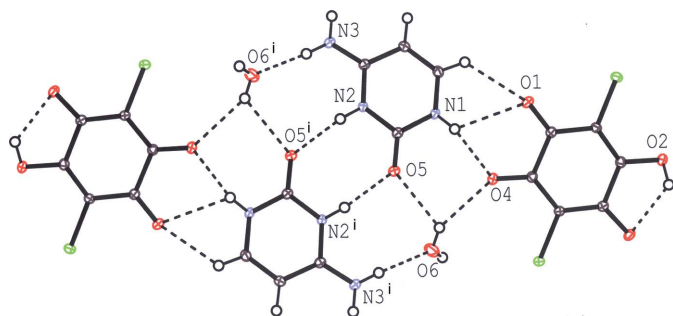
The title compound, (I), was prepared in order to extend our study on $D-\text{H} \cdots A$ hydrogen bonding ($D = \text{N}, \text{O}$ or C ; $A = \text{N}, \text{O}$ or Cl) in the chloranilic acid-pyridine system. Crystal structures have been analyzed for the compounds of diazine (Ishida & Kashino, 1999*a,b,c*), methylpyridine (Ishida & Kashino, 2002; Ishida, 2004*a,b*) and carboxypyridine (Tabuchi *et al.*, 2005; Gotoh *et al.*, 2006). Cytosine, one of the nucleotide building blocks, is a strong base and forms salts with organic acids. As such, several crystal structures have been reported (Tamura *et al.*, 1972; Ohki *et al.*, 1975; Takenaka *et al.*, 1980; Gdaniec *et al.*, 1988, 1989; Balasubramanian *et al.*, 1996; Smith *et al.*, 2005).



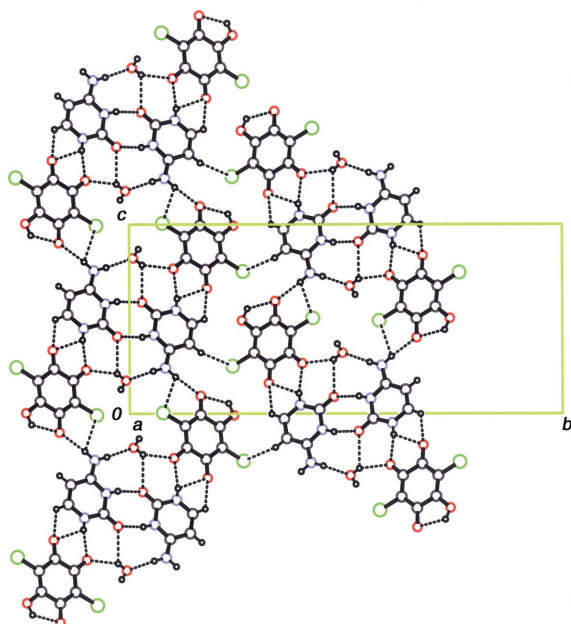
The asymmetric unit in (I) contains a cytosinium cation, a hydrogen chloranilate anion and a water molecule. The three components are held together by bifurcated $\text{N}-\text{H} \cdots \text{O}$ ($\text{N}1-\text{H}2 \cdots \text{O}1$ and $\text{N}1-\text{H}2 \cdots \text{O}4$) and $\text{O}-\text{H} \cdots \text{O}$ ($\text{O}6-\text{H}8 \cdots \text{O}4$ and $\text{O}6-\text{H}8 \cdots \text{O}5$) hydrogen bonds, and a $\text{C}10-\text{H}7 \cdots \text{O}1$ hydrogen bond (Fig. 1 and Table 1). Atom H1 participates only in an intramolecular hydrogen bond ($\text{O}2-\text{H}1 \cdots \text{O}3$). The neighboring units related by an inversion center are connected by $\text{N}2-\text{H}3 \cdots \text{O}5^i$ and $\text{N}3-\text{H}4 \cdots \text{O}6^i$ hydrogen bonds (symmetry codes as in Table 1), resulting in a centrosymmetric dimeric unit (Fig. 2). The dimeric units are further connected by $\text{N}3-\text{H}5 \cdots \text{O}3^{ii}$ and $\text{N}3-\text{H}5 \cdots \text{Cl}2^{ii}$ hydrogen bonds (Table 1), forming a molecular tape running along the

**Figure 1**

The asymmetric unit of (I), with the atom numbering. Displacement ellipsoids of non-H atoms are drawn at the 50% probability level. N—H...O, O—H...O and C—H...O hydrogen bonds are indicated by dashed lines.

**Figure 2**

A view of the dimeric unit of (I), showing the hydrogen-bonding scheme (dashed lines) [symmetry code (i) as given in Table 1].

**Figure 3**

A partial packing diagram, viewed down the *a* axis, showing the hydrogen-bonded tape and the hydrogen-bonding scheme. Dashed lines show N—H...O, O—H...O, C—H...O, N—H...Cl and C—H...Cl hydrogen bonds.

[10 $\bar{1}$] direction. The tapes are stacked along the *a* axis through O6—H9...Cl2ⁱⁱⁱ hydrogen bonds. The neighboring tapes related by an *n*-glide plane to each other are linked weakly through C9—H6...Cl1^{iv} hydrogen bonds (Fig. 3).

Experimental

Crystals were obtained by slow evaporation of a methanol solution (50 ml) of chloranilic acid with cytosine in a 1:1 molar ratio (0.062 and 0.033 g for chloranilic acid and cytosine, respectively).

Crystal data

$C_4H_6N_3O^+ \cdot C_6HCl_2O_4^- \cdot H_2O$
 $M_r = 338.10$
 Monoclinic, $P2_1/n$
 $a = 3.6656$ (1) Å
 $b = 27.8941$ (15) Å
 $c = 12.1499$ (5) Å
 $\beta = 92.3075$ (15)°
 $V = 1241.30$ (9) Å³

$Z = 4$
 $D_x = 1.809$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 0.56$ mm⁻¹
 $T = 100$ (2) K
 Needle, black
 $0.30 \times 0.15 \times 0.10$ mm

Data collection

Rigaku R-Axis RAPID
 diffractometer
 ω scans
 Absorption correction: multi-scan
 (ABSCOR; Higashi, 1995)
 $T_{min} = 0.791$, $T_{max} = 0.946$

14606 measured reflections
 3612 independent reflections
 3126 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.038$
 $\theta_{max} = 30.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.028$
 $wR(F^2) = 0.073$
 $S = 1.05$
 3612 reflections
 227 parameters
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0432P)^2 + 0.179P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} = 0.001$
 $\Delta\rho_{max} = 0.43$ e Å⁻³
 $\Delta\rho_{min} = -0.32$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
N1—H2...O1	0.796 (17)	2.475 (17)	2.9707 (13)	121.6 (15)
N1—H2...O4	0.796 (17)	1.997 (17)	2.7748 (12)	165.6 (17)
N2—H3...O5 ⁱ	0.826 (17)	2.002 (17)	2.7973 (14)	161.4 (16)
N3—H4...O6 ⁱ	0.849 (19)	1.911 (19)	2.7579 (16)	175.6 (18)
N3—H5...O3 ⁱⁱ	0.861 (16)	2.038 (16)	2.8964 (13)	174.7 (16)
N3—H5...Cl2 ⁱⁱ	0.861 (16)	2.830 (16)	3.2278 (10)	110.1 (12)
O2—H1...O3	0.83 (2)	2.03 (2)	2.6084 (12)	126.6 (19)
O6—H8...O4	0.77 (3)	2.32 (3)	3.0171 (15)	151 (2)
O6—H8...O5	0.77 (3)	2.39 (2)	2.8942 (14)	124 (2)
O6—H9...Cl2 ⁱⁱⁱ	0.78 (3)	2.68 (3)	3.4497 (13)	169 (2)
C9—H6...Cl1 ^{iv}	0.945 (16)	2.794 (16)	3.5494 (12)	137.6 (11)
C10—H7...O1	0.921 (16)	2.438 (14)	3.0362 (13)	122.7 (12)

Symmetry codes: (i) $-x+2, -y, -z+1$; (ii) $x+1, y, z-1$; (iii) $x+1, y, z$; (iv) $x+\frac{1}{2}, -y+\frac{1}{2}, z-\frac{1}{2}$.

H atoms were refined without constraints.

Data collection: *PROCESS-AUTO* (Rigaku/MSC, 2004); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC, 2004); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure* and *PLATON* (Spek, 2003).

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